

# Isotope data from USCGC Healy cruise HLY1202 from the Arctic, North of Alaska in 2012 (OA - Canada Basin project)

**Website:** <https://www.bco-dmo.org/dataset/527905>

**Version:** 09 September 2014

**Version Date:** 2014-09-09

## Project

» [Ocean Acidification in the Canada Basin: Roles of Sea Ice](#) (OA - Canada Basin)

## Program

» [Science, Engineering and Education for Sustainability NSF-Wide Investment \(SEES\): Ocean Acidification \(formerly CRI-OA\)](#) (SEES-OA)

Contributors	Affiliation	Role
<a href="#">Onac, Bogdan</a>	University of South Florida (USF)	Principal Investigator
<a href="#">Wynn, Jonathan</a>	University of South Florida (USF)	Principal Investigator
<a href="#">Robbins, Lisa</a>	United States Geological Survey (USGS)	Co-Principal Investigator, Contact
<a href="#">Gegg, Stephen R.</a>	Woods Hole Oceanographic Institution (WHOI BCO-DMO)	BCO-DMO Data Manager

## Table of Contents

- [Dataset Description](#)
  - [Methods & Sampling](#)
  - [Data Processing Description](#)
- [Data Files](#)
- [Parameters](#)
- [Instruments](#)
- [Deployments](#)
- [Project Information](#)
- [Program Information](#)
- [Funding](#)

## Dataset Description

USGS Arctic Ocean Carbon Cruise 2012: Stable isotope Data  
Stable isotope data for seawater samples from the HLY 2012 cruise

For additional information see:

[HLY1202 Cruise Report, Appendix E](#)

[U.S. Geological Survey Data Series 862 - Methods](#)

[USFSI Stable Isotope Lab](#)

## Methods & Sampling

### Isotope Analyses

Seawater samples were collected for stable isotopic analysis by rinsing, then filling 125-mL bottles such that no headspace remained. Samples were fixed with 50 microliters ( $\mu\text{L}$ ) of saturated mercuric chloride, and sealed with Teflon crimp caps containing a thin coating of Apiezon grease. Samples were refrigerated during storage prior to analysis. Analyses were completed at the University of South Florida Department of Geology Stable Isotope Laboratory using a Thermo-Finnigan Delta V 3 kiloelectron (keV) Isotope Ratio Mass Spectrometer coupled to a GasBench II preparation device. Analyses of  $\text{d}18\text{O}$  of  $\text{H}_2\text{O}$  was completed by equilibrating 200

μL of sample with approximately 12 mL headspace of H<sub>2</sub> in septum-capped vials. After equilibration, the isotopic composition of the headspace gas was measured (methods following Epstein and Mayeda, 1953; Prosser and Scrimgeour, 1995). Analyses of δ<sup>13</sup>C of dissolved inorganic carbon (DIC) was completed by injecting 1 mL of sample into an approximately 12-mL vial that was pre-flushed with He and pre-filled with 1 mL of 85 percent H<sub>3</sub>PO<sub>4</sub> (methods following Assayag and others, 2006). The CO<sub>2</sub> produced by this acid-stripping of the DIC was then measured after 24 h of equilibration. All stable isotope data are expressed in the conventional delta (δ) notation:

$$\Delta = [(R_{\text{sample}} - R_{\text{standard}}) / R_{\text{standard}}] \times 1000\text{‰}$$

where R<sub>sample</sub> and R<sub>standard</sub> is the 18O/16O ratio of the sample and standard, respectively, for d18O. The standard used as a reference for the δ scale is VSMOW for H<sub>2</sub>O. Internal standards were used in the calibration to the VSMOW scale (VEEN and HTAMP waters with d18O = -13.17 per mil (‰) and +15.05‰) and PDB scale Carrara Marble and NBS-18 calcite). Analytical precision on these standards was better than 0.15‰ for d18O and 0.2‰ for d13C.

For additional information see:

[HLY1202 Cruise Report, Appendix E](#)

[U.S. Geological Survey Data Series 862 - Methods](#)

[USFSI Stable Isotope Lab](#)

## Data Processing Description

### Isotope Analyses

Seawater samples were collected for stable isotopic analysis by rinsing, then filling 125-mL bottles such that no headspace remained. Samples were fixed with 50 microliters (μL) of saturated mercuric chloride, and sealed with Teflon crimp caps containing a thin coating of Apiezon grease. Samples were refrigerated during storage prior to analysis. Analyses were completed at the University of South Florida Department of Geology Stable Isotope Laboratory using a Thermo-Finnigan Delta V 3 kiloelectron (keV) Isotope Ratio Mass Spectrometer coupled to a GasBench II preparation device. Analyses of d18O of H<sub>2</sub>O was completed by equilibrating 200 μL of sample with approximately 12 mL headspace of H<sub>2</sub> in septum-capped vials. After equilibration, the isotopic composition of the headspace gas was measured (methods following Epstein and Mayeda, 1953; Prosser and Scrimgeour, 1995). Analyses of δ<sup>13</sup>C of dissolved inorganic carbon (DIC) was completed by injecting 1 mL of sample into an approximately 12-mL vial that was pre-flushed with He and pre-filled with 1 mL of 85 percent H<sub>3</sub>PO<sub>4</sub> (methods following Assayag and others, 2006). The CO<sub>2</sub> produced by this acid-stripping of the DIC was then measured after 24 h of equilibration. All stable isotope data are expressed in the conventional delta (δ) notation:

$$\Delta = [(R_{\text{sample}} - R_{\text{standard}}) / R_{\text{standard}}] \times 1000\text{‰}$$

where R<sub>sample</sub> and R<sub>standard</sub> is the 18O/16O ratio of the sample and standard, respectively, for d18O. The standard used as a reference for the δ scale is VSMOW for H<sub>2</sub>O. Internal standards were used in the calibration to the VSMOW scale (VEEN and HTAMP waters with d18O = -13.17 per mil (‰) and +15.05‰) and PDB scale Carrara Marble and NBS-18 calcite). Analytical precision on these standards was better than 0.15‰ for d18O and 0.2‰ for d13C.

The QA/QC process compared analysis results with calibrated reference standards

For additional information see:

[HLY1202 Cruise Report, Appendix E](#)

[U.S. Geological Survey Data Series 862 - Methods](#)

[USFSI Stable Isotope Lab](#)

## BCO-DMO Processing Notes

- Generated from original file: "HLY1202\_Isotopes.csv" contributed by Lisa Robbins
- Parameter names edited to conform to BCO-DMO naming convention found at [Choosing Parameter Name](#)
- Date reformatted from MM/DD/YYYY to YYYYMMDD
- Time reformatted from HH:MM to HHMM
- "nd" (no data) inserted into blank fields

## Data Files

File
<b>Isotopes.csv</b> (Comma Separated Values (.csv), 17.80 KB) MD5:327d2a565fc24821e11871deb0443ec0
Primary data file for dataset ID 527905

## Parameters

Parameter	Description	Units
Sample_ID	Sample identification number. Duplicate samples identified by additional letter qualifier.	text
Bottle_ID	Used to identify rosette bottles from CTD station casts. Deepest bottle is 1. Shallowest bottle is 24.	dimensionless
Latitude	Latitude (South is negative); WGS 84	decimal degrees
Longitude	Longitude (West is negative); WGS 84	decimal degrees
Depth	Depth at which sample was taken. Zero (0) indicates surface sample	meters
d18O	Ratio of Oxygen-18 to Oxygen-16 as compared to VSMOW standard in per mil (parts per thousand).	Per mil (parts per thousand)
d13C	Ratio of Carbon-13 to Carbon-12 as compared to VPDB standard in per mil (parts per thousand).	Per mil (parts per thousand)

## Instruments

<b>Dataset-specific Instrument Name</b>	CTD SBE 911plus
<b>Generic Instrument Name</b>	CTD Sea-Bird SBE 911plus
<b>Dataset-specific Description</b>	Discrete samples from vertical profile casts were collected at 4 locations. A 24-bottle Niskin rosette (12-L bottle volume) with an electronic trigger was fitted with a Sea-Bird SBE 911plus CTD and altimeter.
<b>Generic Instrument Description</b>	The Sea-Bird SBE 911 plus is a type of CTD instrument package for continuous measurement of conductivity, temperature and pressure. The SBE 911 plus includes the SBE 9plus Underwater Unit and the SBE 11plus Deck Unit (for real-time readout using conductive wire) for deployment from a vessel. The combination of the SBE 9 plus and SBE 11 plus is called a SBE 911 plus. The SBE 9 plus uses Sea-Bird's standard modular temperature and conductivity sensors (SBE 3 plus and SBE 4). The SBE 9 plus CTD can be configured with up to eight auxiliary sensors to measure other parameters including dissolved oxygen, pH, turbidity, fluorescence, light (PAR), light transmission, etc.). more information from Sea-Bird Electronics

<b>Dataset-specific Instrument Name</b>	Niskin bottle
<b>Generic Instrument Name</b>	Niskin bottle
<b>Dataset-specific Description</b>	Discrete samples from vertical profile casts were collected at 4 locations. A 24-bottle Niskin rosette (12-L bottle volume) with an electronic trigger was fitted with a Sea-Bird SBE 911plus CTD and altimeter.
<b>Generic Instrument Description</b>	A Niskin bottle (a next generation water sampler based on the Nansen bottle) is a cylindrical, non-metallic water collection device with stoppers at both ends. The bottles can be attached individually on a hydrowire or deployed in 12, 24, or 36 bottle Rosette systems mounted on a frame and combined with a CTD. Niskin bottles are used to collect discrete water samples for a range of measurements including pigments, nutrients, plankton, etc.

<b>Dataset-specific Instrument Name</b>	pump-ship intake
<b>Generic Instrument Name</b>	Pump - Surface Underway Ship Intake
<b>Dataset-specific Description</b>	Samples were numbered incrementally throughout the duration cruises. A "surface" prefix indicates a sample taken from the ship's flow-through system.
<b>Generic Instrument Description</b>	The 'Pump-underway ship intake' system indicates that samples are from the ship's clean water intake pump. This is essentially a surface water sample from a source of uncontaminated near-surface (commonly 3 to 7 m) seawater that can be pumped continuously to shipboard laboratories on research vessels. There is typically a temperature sensor near the intake (known as the hull temperature) to provide measurements that are as close as possible to the ambient water temperature. The flow from the supply is typically directed through continuously logged sensors such as a thermosalinograph and a fluorometer. Water samples are often collected from the underway supply that may also be referred to as the non-toxic supply. Ideally the data contributor has specified the depth in the ship's hull at which the pump is mounted.

<b>Dataset-specific Instrument Name</b>	Thermo-Finnigan Delta V 3 kiloelectron (keV) Isotope Ratio Mass Spectrometer
<b>Generic Instrument Name</b>	Spectrophotometer
<b>Dataset-specific Description</b>	Analyses were completed at the University of South Florida Department of Geology Stable Isotope Laboratory using a Thermo-Finnigan Delta V 3 kiloelectron (keV) Isotope Ratio Mass Spectrometer coupled to a GasBench II preparation device
<b>Generic Instrument Description</b>	An instrument used to measure the relative absorption of electromagnetic radiation of different wavelengths in the near infra-red, visible and ultraviolet wavebands by samples.

[ [table of contents](#) | [back to top](#) ]

## Deployments

HLY1202

<b>Website</b>	<a href="https://www.bco-dmo.org/deployment/523780">https://www.bco-dmo.org/deployment/523780</a>
<b>Platform</b>	USCGC Healy
<b>Report</b>	<a href="http://dmoserv3.who.edu/data_docs/OA_Canada_Basin/HEALY1202_FINAL_CRUISE_REPORT-3.pdf">http://dmoserv3.who.edu/data_docs/OA_Canada_Basin/HEALY1202_FINAL_CRUISE_REPORT-3.pdf</a>
<b>Start Date</b>	2012-08-26
<b>End Date</b>	2012-09-24
<b>Description</b>	Original cruise data are available from the NSF R2R data catalog USCGC Healy Science-Technical Support Summary From August 25 to September 27, 2012, the United States Coast Guard Cutter (USCGC) Healy was part of an Extended Continental Shelf Project to determine the limits of the extended continental shelf in the Arctic. On a non-interference basis, a USGS ocean acidification team participated on the cruise to collect baseline water data in the Arctic. The collection of data extended from coastal waters near Barrow, Alaska, to 83°2'N., - 175°36'W., and southward back to coastal waters near Barrow and on to Dutch Harbor, Alaska. As a consequence, a number of hypotheses were tested and questions asked associated with ocean acidification, including: - What is the saturation state for different parts of the basin? - What factors drive the saturation state in the different parts of the basin? - How does saturation state compare to other regions? - How do the carbon fluxes compare in the different parts of the basin? - What is the buffering capacity of the water (Revelle factor)? - What kind of variability does carbon demonstrate in the Arctic (near shore versus offshore and diurnal)? During the cruise, underway continuous and discrete water samples were collected, and discrete water samples were collected at stations to document the carbonate chemistry of the Arctic waters and quantify the saturation state of seawater with respect to calcium carbonate. These data are critical for providing baseline information in areas where no data have existed prior and will also be used to test existing models and predict future trends.

[ [table of contents](#) | [back to top](#) ]

## Project Information

### Ocean Acidification in the Canada Basin: Roles of Sea Ice (OA - Canada Basin)

**Website:** <http://coastal.er.usgs.gov/ocean-acidification/polar.html>

**Coverage:** Beaufort Sea, Canada Basin

*Extracted from the NSF award abstract:*

The proposed research aims to identify mechanisms of ocean acidification and carbonate undersaturation in the Canada Basin of the Arctic Ocean. The investigators will use a 3-year dataset to address the role of two specific mechanisms involving sea-ice processes: (1) surface water freshening and undersaturation resulting from recent enhanced melting of multi-year sea-ice, and (2) surface water undersaturation resulting from a "carbon-pumping" mechanism driven by brine rejection and carbonate mineral precipitation during increasingly cyclical seasonal sea ice growth and decay. The proposed work would expand understanding of the inorganic carbon cycle, air-sea CO<sub>2</sub> exchange rates and acidification in the Arctic Ocean. Understanding baselines and how they are changing is important for setting realistic parameters for process studies on the effects of ocean acidification on flora and fauna.

Models project the Arctic Ocean will become undersaturated with respect to carbonate minerals in the next decade. Recent field results indicate parts may already be undersaturated in late summer months when ice melt is at its greatest extent. However, few comprehensive datasets of carbonate system parameters in the Arctic Ocean exist. Researchers from the U.S. Geological Survey (USGS) and University of South Florida (USF) collected high-resolution measurements of pCO<sub>2</sub>, pH, total dissolved inorganic carbon (DIC), total alkalinity (TA), and carbonate (CO<sub>3</sub><sup>2-</sup>) from the Canada Basin that fill critical information gaps concerning Arctic carbon variability. A Multiparameter Inorganic Carbon Analyzer (MICA) was used to collect approximately 1,800 measurements of pH and DIC along an 11,965-km trackline in August and September 2012. In addition, over

500 discrete surface water samples were taken. These data are being used to characterize and model regional pCO<sub>2</sub>, pH, and carbonate mineral saturation state. A high-resolution, three-dimensional map of these results will be presented.

Data collected on the August 2012 Arctic Cruise will be used to create regional maps of seawater carbonate parameters, including pCO<sub>2</sub> flux/change maps, and derivative maps on saturation state. Maps depicting pCO<sub>2</sub> and carbonate saturation states over large latitudinal and nearshore to offshore gradients are needed for the Arctic, where significant decline of carbonate ecosystems, habitats, and calcifying organisms are predicted over the next decade. The data will allow the USGS to map variations in ocean chemistry along designated tracks and will be used in models to predict future Arctic Ocean saturation states.

*Note: Jonathan Wynn is a former Principal Investigator for this project*

[ [table of contents](#) | [back to top](#) ]

---

## Program Information

### Science, Engineering and Education for Sustainability NSF-Wide Investment (SEES): Ocean Acidification (formerly CRI-OA) (SEES-OA)

**Website:** [https://www.nsf.gov/funding/pgm\\_summ.jsp?pims\\_id=503477](https://www.nsf.gov/funding/pgm_summ.jsp?pims_id=503477)

**Coverage:** global

NSF Climate Research Investment (CRI) activities that were initiated in 2010 are now included under Science, Engineering and Education for Sustainability NSF-Wide Investment (SEES). SEES is a portfolio of activities that highlights NSF's unique role in helping society address the challenge(s) of achieving sustainability. Detailed information about the SEES program is available from NSF ([https://www.nsf.gov/funding/pgm\\_summ.jsp?pims\\_id=504707](https://www.nsf.gov/funding/pgm_summ.jsp?pims_id=504707)).

In recognition of the need for basic research concerning the nature, extent and impact of ocean acidification on oceanic environments in the past, present and future, the goal of the SEES: OA program is to understand (a) the chemistry and physical chemistry of ocean acidification; (b) how ocean acidification interacts with processes at the organismal level; and (c) how the earth system history informs our understanding of the effects of ocean acidification on the present day and future ocean.

#### Solicitations issued under this program:

[NSF 10-530](#), FY 2010-FY2011

[NSF 12-500](#), FY 2012

[NSF 12-600](#), FY 2013

[NSF 13-586](#), FY 2014

NSF 13-586 was the final solicitation that will be released for this program.

#### PI Meetings:

[1st U.S. Ocean Acidification PI Meeting](#) (March 22-24, 2011, Woods Hole, MA)

[2nd U.S. Ocean Acidification PI Meeting](#) (Sept. 18-20, 2013, Washington, DC)

[3rd U.S. Ocean Acidification PI Meeting](#) (June 9-11, 2015, Woods Hole, MA - Tentative)

#### NSF media releases for the Ocean Acidification Program:

[Press Release 10-186 NSF Awards Grants to Study Effects of Ocean Acidification](#)

[Discovery Blue Mussels "Hang On" Along Rocky Shores: For How Long?](#)

[Discovery nsf.gov - National Science Foundation \(NSF\) Discoveries - Trouble in Paradise: Ocean Acidification This Way Comes - US National Science Foundation \(NSF\)](#)

[Press Release 12-179 nsf.gov - National Science Foundation \(NSF\) News - Ocean Acidification: Finding New Answers Through National Science Foundation Research Grants - US National Science Foundation \(NSF\)](#)

[Press Release 13-102 World Oceans Month Brings Mixed News for Oysters](#)

[Press Release 13-108 nsf.gov - National Science Foundation \(NSF\) News - Natural Underwater Springs Show How Coral Reefs Respond to Ocean Acidification - US National Science Foundation \(NSF\)](#)

[Press Release 13-148 Ocean acidification: Making new discoveries through National Science Foundation research grants](#)

[Press Release 13-148 - Video nsf.gov - News - Video - NSF Ocean Sciences Division Director David Conover answers questions about ocean acidification. - US National Science Foundation \(NSF\)](#)

[Press Release 14-010 nsf.gov - National Science Foundation \(NSF\) News - Palau's coral reefs surprisingly resistant to ocean acidification - US National Science Foundation \(NSF\)](#)

[Press Release 14-116 nsf.gov - National Science Foundation \(NSF\) News - Ocean Acidification: NSF awards \\$11.4 million in new grants to study effects on marine ecosystems - US National Science Foundation \(NSF\)](#)

[ [table of contents](#) | [back to top](#) ]

---

## Funding

Funding Source	Award
<a href="#">NSF Arctic Sciences (NSF ARC)</a>	<a href="#">PLR-1220032</a>

[ [table of contents](#) | [back to top](#) ]