

# Suspended PIC, PC, PN data collected along a North Pacific transect between Hawaii and Alaska on R/V Kilo Moana cruise KM1712 in August 2017

**Website:** <https://www.bco-dmo.org/dataset/860409>

**Data Type:** Cruise Results

**Version:** 1

**Version Date:** 2021-09-21

## Project

» [Ocean Acidification - Collaborative Research: Measuring the kinetics of CaCO<sub>3</sub> dissolution in seawater using novel isotope labeling, laboratory experiments, and in situ experiments](#) (CaCO<sub>3</sub> dissolution)

Contributors	Affiliation	Role
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## Abstract

This dataset includes general measurements for in situ pump casts at 5 stations on a transect between Hawaii and Alaska. Data was collected in August 2017 onboard R/V Kilo Moana cruise KM1712.

## Table of Contents

- [Coverage](#)
- [Dataset Description](#)
  - [Methods & Sampling](#)
  - [Data Processing Description](#)
- [Data Files](#)
- [Related Publications](#)
- [Parameters](#)
- [Instruments](#)
- [Deployments](#)
- [Project Information](#)
- [Funding](#)

## Coverage

**Spatial Extent:** N:50 E:-148 S:23 W:-158

**Temporal Extent:** 2017-08-02 - 2017-08-24

## Dataset Description

North Pacific, 150 W, 20 to 60 N, all depths

## Methods & Sampling

Suspended particles were collected at 5 different stations along a North Pacific transect between Hawaii and Alaska. Samples were collected on 142 mm diameter Glass Fiber Filters (GFF) using McLane Pumps. Particulate

Inorganic Carbon (PIC) content was measured by acidifying a subsample of the GFF filter and measuring total CO<sub>2</sub> released on a Picarro gas concentration analyzer. Total Particulate Carbon (PC) and Particulate Nitrogen (PN) were analyzed by burning a subsample of the GFF on an Elemental Analyzer (EA).

## Data Processing Description

### BCO-DMO processing:

- Converted latitude and longitude to decimal degrees.
- Created column for ISO 8601 formatted times (UTC/GMT timezone)
- Rounded column values to submitter preference
- Modified parameter (column) names to conform with BCO-DMO naming conventions.
  - No spaces, hyphens, commas, parentheses, or Greek letters.
  - The only characters allowed are A-Z, a-z, 0-9, and underscores.

[ [table of contents](#) | [back to top](#) ]

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## Data Files

File
<b>pump_suspended_particles.csv</b> (Comma Separated Values (.csv), 3.55 KB) MD5:b4ddc983fe688e9c1ba0cc2536de56e3
Primary data file for dataset ID 860409

[ [table of contents](#) | [back to top](#) ]

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## Related Publications

Dong, S., Berelson, W. M., Rollins, N. E., Subhas, A. V., Naviaux, J. D., Celestian, A. J., Liu, X., Turaga, N., Kemnitz, N. J., Byrne, R. H., & Adkins, J. F. (2019). Aragonite dissolution kinetics and calcite/aragonite ratios in sinking and suspended particles in the North Pacific. *Earth and Planetary Science Letters*, 515, 1–12.

<https://doi.org/10.1016/j.epsl.2019.03.016>

*Methods*

,

*Results*

[ [table of contents](#) | [back to top](#) ]

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## Parameters

Parameter	Description	Units
Cruise_ID	Cruise identifier	unitless
Cruise_synonym	Cruise name	unitless
Station	Station number	unitless
Longitude	Longitude of sample collection (West is negative)	decimal degrees
Latitude	Latitude of sample collection	decimal degrees
Depth	Depth of sample collection	meters (m)
Deploy_Time_local	Time of pump deployment (local, Hawaii Standard Time)	unitless
Recover_Time_local	Time of pump recovery (local, Hawaii Standard Time)	unitless
d13C	Delta 13C of particulate carbon	per mil
PC	Concentration of total particulate carbon	micrograms per liter (ug/L)
d15N	Delta 15N of particulate nitrogen versus air	per mil
PN	Concentration of total particulate nitrogen	micrograms per liter (ug/L)
PIC_POC_ratio	Ratio of particulate inorganic carbon to particulate organic carbon	unitless
Deploy_DateTime_UTC	Time of pump deployment in ISO8601 format	unitless
Recover_DateTime_UTC	Time of pump recovery in ISO8601 format	unitless

[ [table of contents](#) | [back to top](#) ]

## Instruments

<b>Dataset-specific Instrument Name</b>	Picarro Cavity Ring-Down Spectroscopy Gas Analyzer (G2131-i)
<b>Generic Instrument Name</b>	Cavity enhanced absorption spectrometers
<b>Dataset-specific Description</b>	Particulate Inorganic Carbon (PIC) content was measured by acidifying a subsample of the GFF filter and measuring total CO <sub>2</sub> released on Picarro
<b>Generic Instrument Description</b>	Instruments that illuminate a sample inside an optical cavity, typically using laser light, and measure the concentration or amount of a species in gas phase by absorption spectroscopy. Techniques include cavity ring-down spectroscopy (CRDS) and integrated cavity output spectroscopy (ICOS).

<b>Dataset-specific Instrument Name</b>	Picarro Cavity Ring-Down Spectroscopy Gas Analyzer (G2131-i)
<b>Generic Instrument Name</b>	CO <sub>2</sub> Analyzer
<b>Dataset-specific Description</b>	Particulate Inorganic Carbon (PIC) content was measured by acidifying a subsample of the GFF filter and measuring total CO <sub>2</sub> released on Picarro
<b>Generic Instrument Description</b>	Measures atmospheric carbon dioxide (CO <sub>2</sub> ) concentration.

<b>Dataset-specific Instrument Name</b>	Costech ECS4010 CHNSO Elemental Analyzer
<b>Generic Instrument Name</b>	Elemental Analyzer
<b>Dataset-specific Description</b>	Total Particulate Carbon (PC) and Particulate Nitrogen (PN) were analyzed by burning a subsample of the GFF on Elemental Analyzer (EA).
<b>Generic Instrument Description</b>	Instruments that quantify carbon, nitrogen and sometimes other elements by combusting the sample at very high temperature and assaying the resulting gaseous oxides. Usually used for samples including organic material.

<b>Dataset-specific Instrument Name</b>	McLane Pump
<b>Generic Instrument Name</b>	McLane Pump
<b>Dataset-specific Description</b>	Samples were collected on Glass Fiber Filters using McLane Pumps
<b>Generic Instrument Description</b>	McLane pumps sample large volumes of seawater at depth. They are attached to a wire and lowered to different depths in the ocean. As the water is pumped through the filter, particles suspended in the ocean are collected on the filters. The pumps are then retrieved and the contents of the filters are analyzed in a lab.

[ [table of contents](#) | [back to top](#) ]

## Deployments

### KM1712

<b>Website</b>	<a href="https://www.bco-dmo.org/deployment/837321">https://www.bco-dmo.org/deployment/837321</a>
<b>Platform</b>	R/V Kilo Moana
<b>Start Date</b>	2017-08-01
<b>End Date</b>	2017-09-01
<b>Description</b>	Additional cruise information is available from the Rolling Deck to Repository (R2R): <a href="https://www.rvdata.us/search/cruise/KM1712">https://www.rvdata.us/search/cruise/KM1712</a>

[ [table of contents](#) | [back to top](#) ]

## Project Information

**Ocean Acidification - Collaborative Research: Measuring the kinetics of CaCO<sub>3</sub> dissolution in seawater using novel isotope labeling, laboratory experiments, and in situ experiments (CaCO<sub>3</sub> dissolution)**

**Coverage:** North Pacific, 150 W, 20 to 60 N, all depths

#### *NSF Award Abstract:*

Ocean acidification by anthropogenic carbon dioxide (CO<sub>2</sub>) emissions to the atmosphere will ultimately be balanced by sedimentary carbonate dissolution. The time constant for this reaction, however, is ca. 6,000 years. So, in the coming decades, the ocean's response to CO<sub>2</sub> uptake will be based on the kinetics of supply and removal, not on the thermodynamics of the system. Unfortunately our understanding of the basic rate law for carbonate dissolution in the ocean is lacking. The order of the rate law is still argued to be anywhere from 1 to 4.5; this range represents a major difference in the sensitivity of the system to small changes in saturation state. The relative importance of aragonite vs. calcite dissolution, the influence of magnesium content in the minerals, and the sign of the role of organic matter are all still unknowns in the modern ocean. Of course, a truly useful rate law would be able to combine the relative importance of all of these factors into a predictive rule for how dissolution will respond to ocean acidification.

In this study, researchers at the California Institute of Technology and the University of Southern California will address this problem with a novel set of laboratory and in situ experiments that use carbon-13 (<sup>13</sup>C) tracer labeled biogenic carbonates to measure the dissolution rate under a wide range of saturation states. They will assemble a set of rules that will govern carbonate dissolution in sinking particles and in marine sediments. This will require two sub-projects. First, they will culture several different species of biogenic carbonate producers in the lab under the influence of a strong <sup>13</sup>C label. With enrichments of around 30,000‰ in the calcium carbonate (CaCO<sub>3</sub>), they will measure the change in dissolved inorganic carbon-13 at several time points over 1-2 weeks in specially built high-pressure reaction chambers. The construction of a prototype chamber is completed and it provides the means, for the first time, to control carbonate saturation state by changing seawater chemistry, pressure, and temperature independently. Experiments with pure <sup>13</sup>C labeled inorganic CaCO<sub>3</sub> will provide the inorganic reference frame for the biogenic carbonate results. Secondly, to check the lab-based rate data, they will also use labeled biogenic particles in a simple Niskin bottle based reactor that will be deployable on regular hydrowire. The accumulation of <sup>13</sup>C in the Niskin dissolved inorganic carbon over 1-2 days will provide an initial rate that is directly comparable to the more extensive laboratory study on the same sorts of materials. Using the San Pedro Basin as a test bed for these in situ experiments will sample a range of saturation states in a series of 3-day cruises. This high-sensitivity approach should allow the team to unpack the various components of carbonate dissolution in seawater under rising CO<sub>2</sub> concentrations.

Broader Impacts. Producing a better rate law for carbonate dissolution will have broad implications for the fields of marine chemistry, marine biology, paleoceanography, and for potential societal response to ocean acidification. This rate law sits at the heart of the marine carbonate cycle. In addition, this work will benefit at least two graduate students and promote US-Israel collaborations via the inclusion of Jonathan Erez and his students. The specific involvement of underrepresented high school students in scientific/oceanographic research is built into the efforts of this project as well as ongoing efforts by both PIs to communicate their science to a broad array of non-scientific audiences.

[ [table of contents](#) | [back to top](#) ]

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## Funding

Funding Source	Award
<a href="#">NSF Division of Ocean Sciences (NSF OCE)</a>	<a href="#">OCE-1220600</a>
<a href="#">NSF Division of Ocean Sciences (NSF OCE)</a>	<a href="#">OCE-1220302</a>

[ [table of contents](#) | [back to top](#) ]